1,3-BENZODITHIOLIUM ION MEDIATED ANNULATIONS

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<u>Summary</u>: A mild and selective method for cyclizations onto aromatic systems using 1,3-benzodithiolium ions is described.

The Friedel-Crafts reaction continues to play an important role in annulation chemistry. 1 The utility of this reaction, however, is sometimes limited by the strongly acidic reaction conditions which are often employed. 1a Recently, a number of novel methods have been published which address this problem. 1c,d

We wish to report the results of a study in connection with another project which features a particularly mild cyclization protocol mediated by the 1,3-benzodithiolium carbocation. Typically, several crystals (10 mg) of p-toluenesulfonic acid are added to a solution of ketene dithioacetal $1a-f^4$ (2.0 mmol) in dry acetonitrile (10 ml). The

resulting solution is stored at 0° C for approximately 24 hr at which time the crystals of benzodithiole 2a-f are filtered off. (see Table I)

TABLE I

<u>Ketenedithioacetal</u>	Benzodithiole, a yield(mp)	Tetralone, a yield(mp)	Tetralin, a yield
la) R ₂ ,R ₃ =OMe	76%(141-142°C) ^C	60%(98-100°C)lit ^{6a} 99-100°C	69%(55-57°C)
1b) R_3 , R_{Δ} = OMe	56%(132°-133°C)	55%(104°C)1it ^{6b} 104-105°C	75%
1c) R ₂ , R ₃ =-OCH ₂ -	75%(135°-136°C) ^C	55%(77°C)	91%(41-45°C)
1d) R ₃ =OMe	77%(130°-131°C) ^c	62%(76-78°C)lit ^{6c} 77-79°C	95%
le) R ₁ -R ₄ =H	70% ^d	62% ^b	-
lf) R ₂ =OMe	74%(95°-97°C) ^d	56%(61°C)1it ^{6c} 61-63°C	-
1g) R_1 , R_4 =OMe	-	-	-

a) The $^{\rm I}$ H NMR, CMR, IR, and MS of these compounds were completely consistent with the assigned structure. b) $^{\rm I}$ H NMR and IR were identical to those of the authentic material c) None of the other possible regioisomer was observed. d) Approximately twice the normal quantity of p-TsOH was required for efficient cyclization.

The benzodithioles obtained from this exceptionally mild process are versatile intermediates which can be hydrolyzed to the corresponding tetralones $\underline{3a-f}$ (HgO/BF $_3$ ·Et $_2$ O/aqTHF) 5 or desulfurized with Raney-Nickel (W-2) to the related tetralins $\underline{4a-d}$.

The regioselectivity exhibited by this reaction is particularly noteworthy since only one of two possible isomers was obtained with ketenedithioacetals 1a.c.d. In addition, entry 1g did not yield any cyclized product even under forcing conditions. This selectivity for bond formation remote to ortho substituents is probably steric in nature. It is also interesting that the relatively electron poor system <a href="le-c.c.decline-c

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- a) A. J. Birch, G. S. R. Subba Rao, Aust. J. Chem., 23, 547 (1970); b) G. H. Posner,
 M. J. Chapdelaine, C. M. Lentz, J. Org. Chem., 44, 3661 (1979); c) B. M. Trost,
 M. Reiffen, M. Crimmin, J. Am. Chem. Soc., 101, 257 (1979); d) Y. Tamura, H.-D. Choi,
 H. Shindo, J. Uenishi, H. Ishibashi, Tetrahedron Lett., 81 (1981).
- 2. W. R. H. Hurtley, S. Smiles, J. Chem. Soc., 2263 (1926).
- For recent related chemistry of 1,3-benzodithiolium ions, see: a) S. Ncube,
 A. Pelter, K. Smith, P. Blatcher, S. Warren, Tetrahedron Lett., 2345 (1978);
 b) I. Degani, R. Fochi, J. Chem. Soc., Perkin Trans. I, 1886 (1976);
 c) J. Nakayama,
 K. Fujiwara, M. Hoshino, Bull. Chem. Soc., Japan, 49, 3567 (1976).
- 4. These ketenedithioacetals are prepared in excellent yields (95%-98%) from the corresponding aldehyde⁷ and 2-lithio-2-diethoxy phosphinyl-1,3-benzodithiole:
 K. Akiba, K. Ishikawa, N. Inamoto, Bull. Chem. Soc., Japan, 51, 2674 (1978).
- 5. E. Vedejs, P. Fuchs, J. Org. Chem., 36, 366 (1971).
- a) R. I. Thrift, J. Chem. Soc., (C) 288 (1967); b) N. F. Elmore, T. J. King,
 Ibid, 4425 (1961); c) Aldrich Handbook, 1982.
- 7. The aldehyde precursors are prepared from the corresponding benzaldehyde and the anion of triethyl phosphonoacetate followed by reduction with lithium aluminum hydride and pyridinium chlorochromate oxidation.

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